BASIC THERMODYNAMICS

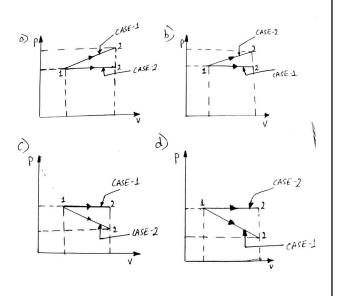
4.

1. For a thermodynamic system to be in thermodynamic equilibium, which of the following can be non uniform through out the system

- (a) temperature
- (b) pressue
- (c) chemical composition
- (d) none of these
- 2. A piston cylinder device with a set of stops initially contains 0.3 kg of water vapour at 1 MPa and 400°C. The location of the stops corresponds to 60% of the initial volume. Now as the water vapour is cooled keeping the pressure constant. The compression work required will be

kJ [Take R for water vapour as 0.4615 kJ/kgK] [Consider water vapour as an ideal gas]

3. A spherical balloon of 1 m diam eter contains a gas at 150 kPa, the gas inside the balloon is heated. Until pressure reaches450 kPa. During the processof heating, the presure inside the balloon is di rectly proportional to the diameter of the balloon. The work done for the process is ______ kJ A piston cylinder device contains a gas. Heat is added to the gas un til the volume becomes double.
In case-I Piston is free to move and in case-2 a linear spring isjust touching the piston before heat addition. After the a heat addition the p-v curve of the



5. During an expansion process, the pressure of gas changes from 100 to 700 kPa according to the rela

tion [P = aV + b], where

 $a = 1220 \left[\frac{kPa}{m^3} \right]$ and b is a constant. If the initial volume of gas is 0.2 m³, The work done dur ing the proces will be _____kJ A mixture of gases expands at constant pressure from 1 MPa,

0.03 m³ to 0.06m³ with 84 kJ positive heat transfer. There is no work other than done on the pis

6.

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